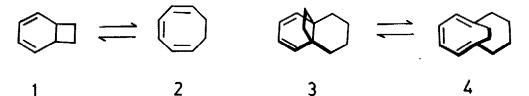
ON THE FLUXIONAL BEHAVIOUR OF A POLYCYCLIC (4.4.2) PROPELLA-2,4-DIENE

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Summary: Existence of bicyclo (4.2.0) octa-2,4-diene \longrightarrow 1,3,5-cyclooctatriene type of valence isomerisation in a polycyclic propellane framework has been demonstrated through cycloaddition and thermolysis reactions.

Although the symmetry allowed bicyclo (4.2.0) octa-2,4-diene $\underline{1} \rightleftharpoons 1,3,5$ cyclooctatriene $\underline{2}$ valence isomerisation has been extensively observed^{1a,b}, attempts to detect such an equilibrium in the propellane framework e.g., $\underline{3} \rightleftharpoons 4$ and congeners have not proved fruitful^{1b-f}. We have recently reported² a novel acid catalysed rearrangement of the hexacyclic dione $\underline{5}^3$, embodying the

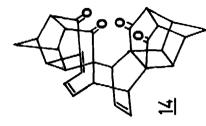


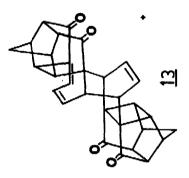
propellanic framework of 3, to the novel dimer <u>14</u> and implicated the valence tautomer <u>6</u> as an intermediate. This observation provided the stimulus towards a direct search for the <u>5</u> \leftarrow <u>6</u> type of equilibrium process. In this communication, we report on the <u>5</u> \leftarrow <u>6</u> valence isomerisation as unravelled through cycloaddition reactions and thermal activation.

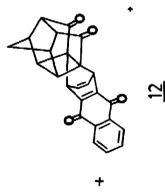
Previously, we have reported² that <u>5</u> exhibited appreciable thermal stability⁴ and in conformity with this observation its ¹H nmr spectrum



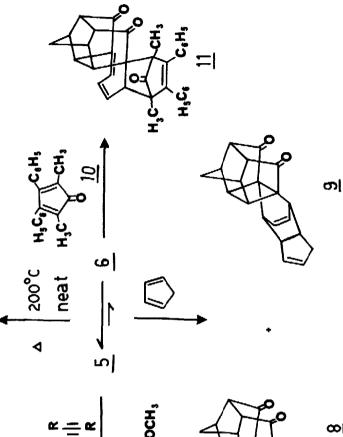
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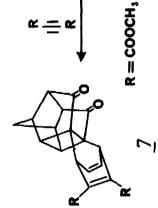














remained temperature invariant (+30 to +130°, DMSO-d₆, 2hr equilibration time) and gave no discernible clue to the presence of triene form 6. However, the response of 5 to cycloaddition reactions proved to be incisive and decisive. While dienophiles like dimethyl acetylenic ester reacted³ with 5 to capture its 1,3-cyclohexadiene moiety and furnished $\pi^4 s^+ \pi^2 s$ adduct 7, mp.186-7°, in high yield; the reactive 1,3-dienes intercepted the highly strained, dienophilic valence tautomer <u>6</u> and furnished novel $\pi^4 s^+ \pi^2 s^-$ adducts. Thus, reaction of 5 with 1,3-cyclopentadiene (Cp) in refluxing toluene (7hr) and chromatography (SiO,gel) furnished two crystalline 1:1 adducts in 80 and 4% yield. The major adduct, mp.138-39°, is formulated as 8 on the basis of spectral data: $C_{20}^{H_{18}0_{2}}$, UV: $\lambda_{max}^{MeoH_{283}}$ nm ($\epsilon = -2000$) and 240 ($\epsilon = -5000$); ir (KBr): 1736 (strong, cyclopentanone), 1640 cm⁻¹ (medium, conjugated carbonyl); ¹H nmr (CDC1₂):**5**6.92 (1H, br s), 6.4 (1H,m), 6.28 (1H,m), 6.1 (1H,d with st, J=10Hz), 5.28 (1H, dd, $J_1 = 10$ Hz, $J_2 = 6.5$ Hz); ¹³C nmr (CDCl₃): δ 218.7(s), 203.4(s), 146.7 (s), 139.3(d), 139.1(d), 137.5(d), 135.4(d), 130.8(d), 72.7(s), 58.9(d), 55.7 (d), 54.1(d), 52.5(d), 47.3(d), 46.1(d), 44.6(?), 44.5(?), 44.4(d), 44.2(t), $35.1(t)^5$. In particular the presence of six olefinic C's, including a fully substituted one at $\pmb{\delta}$ 146.7, and the presence of widely separated carbonyls C's at 218.7 and 203.4 in the ¹³C nmr provided convincing evidence for the structural assignment 8. The minor adduct 9, mp.258-59°, was readily formulated as the Diels-Alder adduct of 5 and Cp with the later participating as the 2% component. In an analogous manner, reaction of hemicyclone 10 with 5 in refluxing toluene (68hr) furnished the 1:1 adduct 11, mp.227-28° in $\sim 20\%$ yield $^{5-7}$.

When subjected to forcing conditions⁴ of thermal activation (200°+10°, neat, 2hr) 5 yielded a complex mixture of products from which four products 9, 12, 13 & 14 could be isolated in 1:3:1:2 ratio through SiO₂ gel chromatography. Structure of 9 was quickly recognised as it was found identical with the minor product obtained in the reaction of Cp with 5. Similarly, 14 could be readily identified as the dimer obtained by us earlier². The structure of <u>12</u> was indicated by its spectral characteristics 5,6 and further verified through its preparation via dehydrogenation of the Diels-Alder adduct of naphthoquinone and 5. Finally, the structure of the dimer 13 was tentatively deduced from the similarity of its spectral data (¹³C nmr and ¹H nmr)^{5,6} with the dimer 14 of firmly established stereostructure. Formation of products 9, 12, 13 & 14 in the thermolysis of 5 indicates that both the thermal fragmentation¹f and valence isomerisation are operative under our reaction conditions. On one hand, sequential 2+2 and 4+2 cycloreversion of 5 generates cyclopentadiene and naphthoquinone which re-add in $\pi^4 + \pi^2 + \pi^2$ fashion to 5 and furnish adducts 9 and 12. The other pathway from 5 leads to the strained, reactive valence isomer 6 and is further captured by 5 to form dimers 13 and 14.

<u>Acknowledgement</u>: We would like to thank the University Grants Commission for the financial support and Mr.A.Srikrishna for some preliminary experiments. 1372

References and Notes:

1(a) R.Huisgen, G.Boche, A.Dahmen and W.Hechtl, Tetrahedron Letts., 5215(1968); (b) D.Ginsburg, MTP International Review of Science, Organic Chemistry Series 2, Vol.5, Butterworths, London (1976); (c) L.A.Paquette, J.C.Phillips & R.E.Wingard jr, J.Amer.Chem.Soc., 93, 4516 (1971); (d) E.Vogel, E.Maier and J.Eimer, Tetrahedron Letts., 655 (1966); (e) J.J. Bloomfield and J.R.S.Irelan, ibid., 2971 (1966), (f) L.A.Paquette and G.L.Thomson, J.Amer.Chem.Soc., 94, 7127 (1972). 2. G.Mehta, V.K.Singh, A.Srikrishna, T.S.Cameron & C.Chan, Tetrahedron Letts., 4595 (1979). 3. Hexacyclo (7.4.2.0^{1,9}.0^{3,7}.0^{4,14}.0^{6,15}) pentadeca-10,12-diene-2,8-dione 5 was prepared as reported by A.S.Kushner, Tetrahedron Letts., 3275 (1971). 4. In refluxing o-dichlorobenzene, 5 remained essentially unchanged for few hrs but prolonged reaction time led to complex mixture of products. 5. 1 H nmr and 13 C nmr spectra were recorded on a JEOL MH-100 and FX-100 instruments at 100 MHz and 25.0 MHz, respectively. Chemical shifts are with respect to internal TMS. Off-resonance multiplicities, when clearly discernible, in the ¹³C nmr are given in parenthesis. Many of the ¹H nmr spectral assignments were confirmed by appropriate spin decoupling and LIS studies. 6. <u>Compound 9</u>: C₂₀H₁₈O₂, ir(KBr): 1725, 1740 cm⁻¹ (cyclopentanone); ¹H nmr(CDCL): 6.34(2H,t,J=3.5Hz), 5.52(2H,m), 3.3(1H,m), 1.62-3.10(13H,m): ¹³C nmr (CDCl₂):δ 212.6(s), 133.9(d), 132.5(d), 131.4(2c,d), 55.8(d), 45.9(d), 43.4(d), 41.9(d), 40.8(?), 38.9(s), 36.6(d), 34.6(d), 33.4(t). Some of the carbon resonances seem to be overlapping. Compound <u>11</u>: $C_{34}H_{28}O_3$, ir(KBr): 1778(Bridge carbonyl), 1738(cyclopentanone), 1630(conjugated carbonyl); ¹H nmr(CDCl₃): **§**6.8-7.4(11H,

m), 6.26(1H, d, J=10Hz), 5.56(1H, dd, J₁=10Hz, J₂=6Hz), 3.3(3H, br s), 2.9(2H, m), 2.76(2H, m), 1.98(2H, br s), 1.34(3H, s), 1.16(3H, s). <u>Compound 12</u>: mp. > 300°, $C_{25}H_{16}O_{4}^{0}$, UV: \bigwedge_{max}^{MeOH} 330,270,250,244 & 240 nm; ir(KBr): 1736(cyclopentanone), 1672 cm (conjugated carbonyl); ¹H nmr (TFA): §7.98(2H, m), 7.66(2H, m), 6.7(2H, t,J=3.5Hz), 4.56 (2H, t,J=3.5Hz), 3.14 (2H, m), 2.94 (2H, m), 2.82 (2H, br s), 2.1 (2H, br s). <u>Compound 13</u>: mp. > 300°, $C_{30}H_{24}O_{4}$, UV: \bigwedge_{max}^{MeOH} 276 (\pounds ~ 2600) and 241 nm (\pounds ~ 4200); ir(KBr): 1730, 1740 (cyclopentanone), 1645 cm⁻¹ (conjugated carbonyl); ¹H nmr (CDCl₃): § 6.84 (1H, br s), 6.15-6.7 (3H, m), 5.84(1H, dd, J₁=6.5, J₂=10Hz), 1.3-3.3, (19 H, m); ¹³C nmr (CDCl₃): § 218.7(s), 212.8(s), 212.4(s), 203.7(s), 145.1(s), 135.5(d), 134.7(d), 133.9(d), 133.2(d), 131.0(d), 68.3(s), 59.1(d), 56.1(d), 55.3(s), 55.3(d), 55.0(d), 51.1(d), 50.6(d), 46.4, 45.4, 45.3, 44.6, 43.6, 43.4, 41.9, 40.8, 35.3, 34.6, 34.2, 32.0(t).

7. We have not ventured to make any stereochemical assignment to adducts $\underline{8}$, $\underline{9}$, $\underline{11} \& \underline{12}$ as it was neither readily deducible from the spectral data nor quite germane to the main theme of this report.

(Received in UK 11 February 1980)